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RE-3551-52

M. Sc. (Part - II) Examination

April / May - 2010

Organic Chemistry : Paper - I

(Natural & Advanced Organic Chemistry)

Time : 3 Hours]

[Total Marks : 70

RE-3551

Instructions :

(1)

नीचे दशांशके निशानीवाणी विगतो उत्तरवही पर अवश्य लपवी. Fillup strictly the details of signs on your answer book.	Seat No. :
Name of the Examination :	<input type="text"/>
<input type="text" value="M. SC. (PART - 2)"/>	<input type="text"/>
Name of the Subject :	<input type="text"/>
<input type="text" value="ORGANIC CHEMISTRY - 1"/>	<input type="text"/>
Subject Code No. : <input type="text" value="3"/> <input type="text" value="5"/> <input type="text" value="5"/> <input type="text" value="1"/>	<input type="text" value="Student's Signature"/>
Section No. (1, 2,.....) : <input type="text" value="1"/>	

- (2) Answer to the two sections should be written in separate answer books.
- (3) Figures to the right indicate full marks of the questions.

- 1 (a) How is it established that Haemin contains vinyl group and propionic acid residue? **12**
- (b) What is meant by bile pigment? Prove the constitution of Billirubin by analytical evidences.
- (c) Discuss the position of hydroxyl, methoxy and carboxyl group in reserpic acid giving suitable reaction.

OR

- 1 (a) Prove the presence of $>N-CH_3$ group and ether linkage in morphine. **12**
- (b) Give the synthesis of unsymmetrical dipyrrol methane and cryptopyrrole.
- (c) Discuss all the degradation products of chlorine-e and give the skeleton structure of chlorine-e.

- 2 (a) Discuss the position of angular methyl groups in cholesterol. 12
- (b) Prove that bile acids are hydroxyl derivatives of 5- β -cholanic acid or 5- α -cholanic acid. Give the synthesis of Diel's Hydrocarbon.
- (c) What are adrenocortical hormones? Explain their importance. Give structures of four such hormones.

OR

- 2 (a) "Cholesterol has ethylenic double bond either between C₄ and C₅ or C₅ & C₆ while hydroxyl group at C₃". Justify 12
- (b) Give the evidences for the position of phenolic -OH group and benzenoid ring in Oestrone.
- (c) What are sex hormones? Classify them giving one example of each. Give the synthesis of Testosterone.
- 3 (a) Alkaline solution of Riboflavin when exposed to light yields Lumilactoflavin. Discuss the structure of Lumilactoflavin. 11
- (b) Discuss the biochemical importance of Vitamin B₁₂.
- (c) Discuss Ruzicka's work to confirm the position of carboxyl group and angular methyl group in abietic acid.

OR

- 3 (a) Give the evidences for the nature of nucleus and position of double bond in zingiberene. 11
- (b) Prove that vitamin K₁ is 2-methyl-3-phytyl-1,4-naphthaquinone.
- (c) Give the synthesis of eudalene.

RE-3552

Instructions :

(1)

नीचे दशांशिक निशानीवाणी विगतो उत्तरवडी पर अवश्य कभवी. Fillup strictly the details of signs on your answer book.	Seat No. :
Name of the Examination :	<input type="text"/> <input type="text"/> <input type="text"/> <input type="text"/> <input type="text"/> <input type="text"/>
<input type="text" value="M. SC. (PART - 2)"/>	<input type="text" value="Student's Signature"/>
Name of the Subject :	
<input type="text" value="ORGANIC CHEMISTRY - 1"/>	
Subject Code No. : <input type="text" value="3"/> <input type="text" value="5"/> <input type="text" value="5"/> <input type="text" value="2"/>	Section No. (1, 2,.....) : <input type="text" value="2"/>

(2) Answer to the two sections should be written in separate answer books.

(3) Figures to the right indicate full marks of the questions.

4 (a) Discuss the principles and concept of green chemistry. 12

(b) Explain the following aqueous phase reaction :

(i) Baeyer-Villager oxidation

(ii) Knoevenagel condensation.

(c) Give the manufacture and uses of the following using greener technology :

(i) Paracetamol

(ii) Adipic acid.

OR

4 (a) What are ionic liquids? Classify them and discuss their applications in Heck reaction. 12

(b) Explain the following aqueous phase reaction :

(i) Claisen-Schmidt condensation

(ii) Diels-Alder reaction.

(c) Give the manufacture and uses of the following using greener technology :

(i) Methyl methacrylate

(ii) Catechol

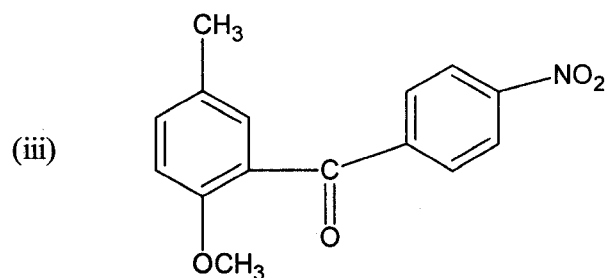
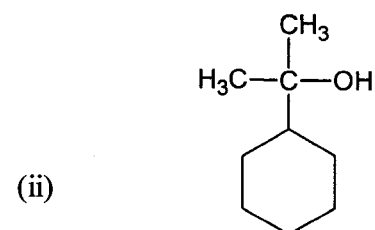
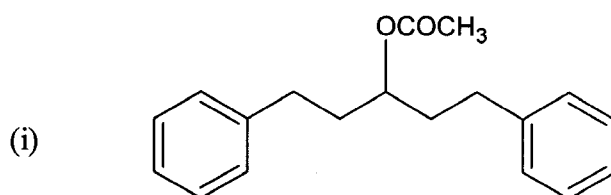
5 (a) Explain the following transformation using appropriate reagents : 12

(i) p-bromo phenol \rightarrow p-hydroxy benzoic acid

(ii) Methyl-3-oxocyclohexane carboxylate \rightarrow
3-(1-hydroxyl-1-methyl ethyl) cyclohexanone.

(iii) Cyclopentanone \rightarrow 1-hydroxy-1-acetyl cyclopentane

(b) Give the disconnection and plan the synthesis for the following molecules :



OR

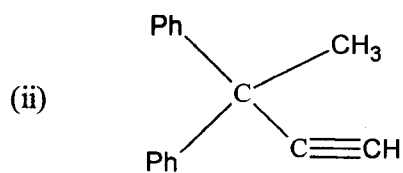
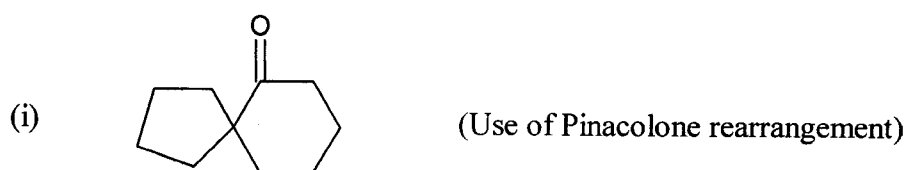
5 (a) Explain the following transformation using appropriate reagents : 12

(i) p-bromo acetophenone \rightarrow p-acetyl benzyl alcohol

(ii) cyclohexanone \rightarrow 1-hydroxy-1-acetyl cyclohexane

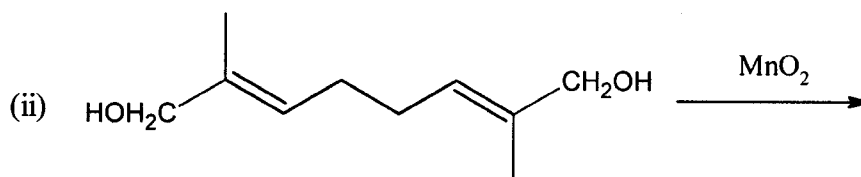
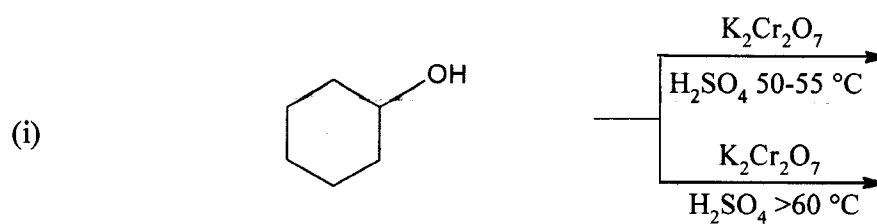
(ii) Valine + Glycine \rightarrow Valinylglycine

(b) Give the disconnection and plan the synthesis for the following molecules :

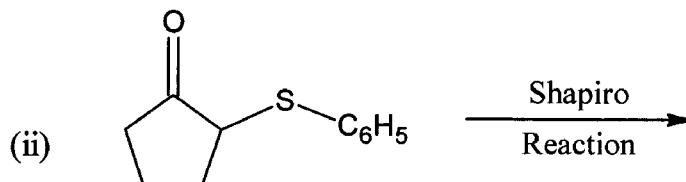
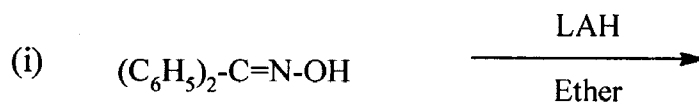


(iii) $\text{Ph-CH}_2\text{-CH(COOEt)}_2$

6 (a) Complete the following reactions and give their mechanism : 11



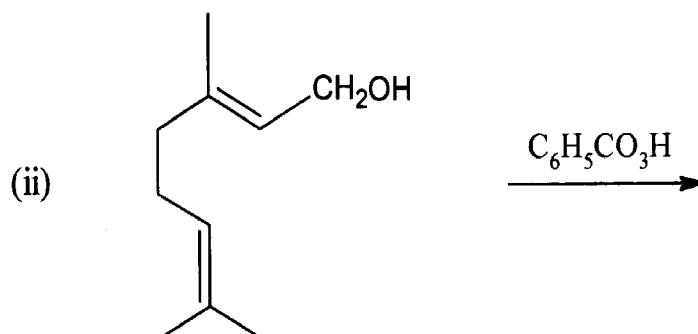
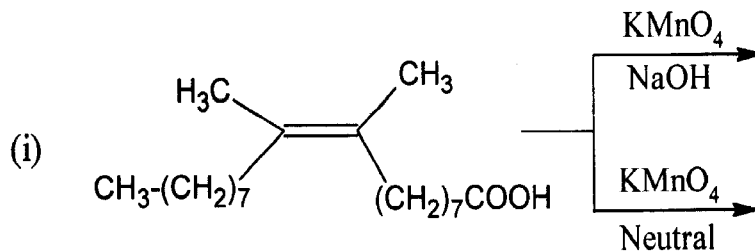
(b) Complete the following reactions and give their mechanism :



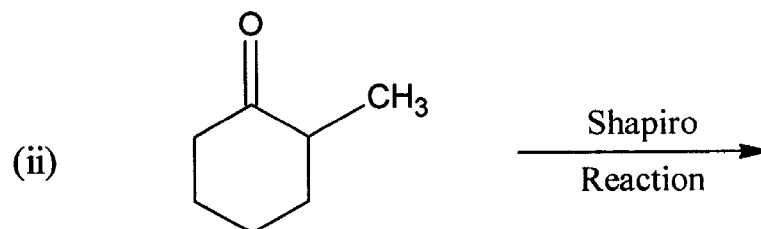
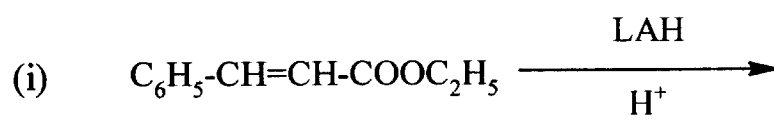
(c) Discuss the Houghton's tea bag procedure of parallel synthesis and give the limitation of combinatorial synthesis.

OR

6 (a) Complete the following reactions and give their mechanism: 11



(b) Complete the following reactions and give their mechanism :



(c) Define the term combi-chem. Discuss the solid phase techniques and the anchor-linker concept.
